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2-Chloro-4,6-bis(1,1,3,3-tetramethylbutylamino)-s-triazine

Fan-Yong Yan

College of Materials and Chemical Engineering, Tianjin Polytechnic University,
Tianjin 300072, People's Republic of China
Correspondence e-mail: dengyi2004@126.com

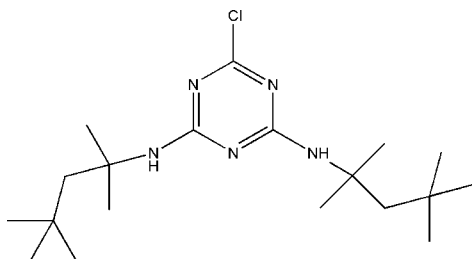
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Key indicators: single-crystal X-ray study; $T = 294$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.049; wR factor = 0.163; data-to-parameter ratio = 13.8.

In the title compound, $\text{C}_{19}\text{H}_{36}\text{ClN}_5$, intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into zigzag chains.

Related literature

For general background, see Borzatta & Carrozza (1991). For related structures, see Deng *et al.* (2006). For related literature, see: Kaiser & Thurston (1951).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{36}\text{ClN}_5$
 $M_r = 369.98$
Orthorhombic, *Pbca*

$a = 11.9729$ (18) Å
 $b = 16.584$ (2) Å
 $c = 22.581$ (4) Å

$V = 4483.6$ (12) Å³
 $Z = 8$
Mo $K\alpha$ radiation

$\mu = 0.18$ mm⁻¹
 $T = 294$ (2) K
 $0.22 \times 0.18 \times 0.10$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.961$, $T_{\max} = 0.982$
21444 measured reflections
3904 independent reflections
1999 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.086$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.163$
 $S = 1.04$
3904 reflections
282 parameters
120 restraints
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.21$ e Å⁻³
 $\Delta\rho_{\min} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{N4}-\text{H4}\cdots\text{N3}^{\text{i}}$ | 0.86 | 2.39 | 3.245 (4) | 174 |
| $\text{N5}-\text{H5}\cdots\text{N2}^{\text{ii}}$ | 0.86 | 2.29 | 3.147 (3) | 179 |

Symmetry codes: (i) $x - \frac{1}{2}, y, -z + \frac{1}{2}$; (ii) $x + \frac{1}{2}, y, -z + \frac{1}{2}$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RN2026).

References

- Borzatta, V. & Carrozza, P. (1991). Eur. Patent EP 0 462 069.
Bruker (1997). *SMART*, *SAINTE* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
Deng, Y., Wang, X.-J., Wen, F., Wang, L. & Zhang, Y. (2006). *Acta Cryst.* **E62**, o5207–o5208.
Kaiser, D. W. & Thurston, J. T. (1951). *J. Am. Chem. Soc.* **73**, 2984–2986.
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Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

supplementary materials

Acta Cryst. (2007). E63, o3327 [doi:10.1107/S1600536807030334]

2-Chloro-4,6-bis(1,1,3,3-tetramethylbutylamino)-s-triazine

F.-Y. Yan

Comment

The title compound, (I), has attracted much attention as an important intermediate in the synthesis of hindered light stabilizers (Borzatta & Carrozza, 1991). These triazine ring containing compounds are widely used (Deng *et al.*, 2006). Herein we report the crystal structure of the title compound (Fig. 1).

The triazine ring in (I) is essentially planar despite having two substituents. The r.m.s. deviation from the mean plane is 0.009 (7) Å.

Intermolecular N—H···N hydrogen bonds are observed which link the molecules into zigzag chains.

Experimental

The title compound was prepared according to the method of Kaiser & Thurston (1951). 2,4,6-Trichloro-1,3,5-triazine (40.0 g, 0.217 mol) was dissolved in toluene (120 ml) and then cooled to 278 K. With stirring, a solution of 2,4,4-trimethylpentan-2-amine (27.5 g, 0.213 mol) in toluene (50 ml) was then added dropwise to the mixture over a period of 0.5 h. A solution of Na₂CO₃ (23.02 g, 0.217 mol) in water (50 ml) was then added dropwise for 0.5 h. The mixture was stirred at 273–278 K for a further 3 h. Another portion of 2,4,4-trimethylpentan-2-amine (27.5 g, 0.213 mol) and solid Na₂CO₃ (23.02 g, 0.217 mol) were added to the mixture, maintaining the temperature at 338 K for 5 h. The organic layer was washed with water and then concentrated *in vacuo*. The title compound (67.9 g) was obtained in powder form in a yield of 85.5%. Crystals of (I) were obtained by slow evaporation of a solution of methanol (m.p. 427–429 K).

Refinement

All H atoms were constrained; positioned geometrically (C—H=0.96–0.97 Å) and refined as riding with $U_{\text{iso}}(\text{H})=1.2U_{\text{eq}}(\text{carrier})$ or $1.5U_{\text{eq}}(\text{methyl groups})$. Atoms C9, C10 and C11 are disordered over two conformations with an occupancy ratio of 0.61 (3):0.39 (3) and atoms C17, C18 and C19 are disordered over two conformations in a 0.758 (9):0.242 (9) ratio.

Figures

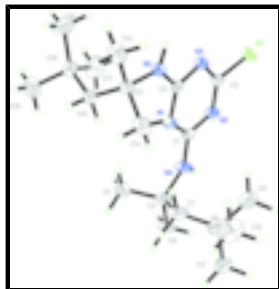


Fig. 1. A view of the molecule (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii. The disordered methyl groups are omitted for clarity.

2-Chloro-4,6-bis(1,1,3,3-tetramethylbutylamino)-s-triazine

Crystal data

$C_{19}H_{36}ClN_5$

$M_r = 369.98$

Orthorhombic, *Pbca*

$a = 11.9729$ (18) Å

$b = 16.584$ (2) Å

$c = 22.581$ (4) Å

$V = 4483.6$ (12) Å³

$Z = 8$

$F_{000} = 1616$

$D_x = 1.096$ Mg m⁻³

Melting point: 154-156 K

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 2903 reflections

$\theta = 2.3$ – 21.2°

$\mu = 0.18$ mm⁻¹

$T = 294$ (2) K

Block, colorless

$0.22 \times 0.18 \times 0.10$ mm

Data collection

Bruker SMART CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 294$ (2) K

φ and ω scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

$T_{\min} = 0.961$, $T_{\max} = 0.982$

21444 measured reflections

3904 independent reflections

1999 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.086$

$\theta_{\text{max}} = 25.0^\circ$

$\theta_{\text{min}} = 1.8^\circ$

$h = -10 \rightarrow 14$

$k = -16 \rightarrow 19$

$l = -26 \rightarrow 26$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.049$

$wR(F^2) = 0.163$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0711P)^2 + 1.3408P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$S = 1.04$ $(\Delta/\sigma)_{\max} = 0.001$
 3904 reflections $\Delta\rho_{\max} = 0.21 \text{ e } \text{\AA}^{-3}$
 282 parameters $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$
 120 restraints Extinction correction: none
 Primary atom site location: structure-invariant direct methods

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|-----|--------------|--------------|--------------|----------------------------------|-----------|
| C11 | 0.06499 (7) | 0.22660 (5) | 0.33227 (4) | 0.0512 (3) | |
| N1 | 0.0743 (2) | 0.08356 (15) | 0.16893 (10) | 0.0354 (6) | |
| N3 | 0.1692 (2) | 0.14722 (15) | 0.25042 (11) | 0.0367 (6) | |
| N2 | -0.0295 (2) | 0.14834 (16) | 0.24687 (11) | 0.0384 (7) | |
| N4 | -0.1201 (2) | 0.08910 (16) | 0.16998 (11) | 0.0438 (7) | |
| H4 | -0.1780 | 0.1006 | 0.1908 | 0.053* | |
| N5 | 0.2677 (2) | 0.08484 (14) | 0.17770 (11) | 0.0365 (6) | |
| H5 | 0.3233 | 0.1024 | 0.1981 | 0.044* | |
| C1 | -0.0220 (3) | 0.10610 (18) | 0.19509 (13) | 0.0349 (7) | |
| C2 | 0.0689 (3) | 0.16651 (17) | 0.26901 (13) | 0.0347 (7) | |
| C3 | 0.1671 (2) | 0.10456 (17) | 0.19841 (13) | 0.0328 (7) | |
| C4 | -0.1430 (3) | 0.0532 (2) | 0.11131 (15) | 0.0491 (9) | |
| C5 | -0.0922 (3) | -0.0317 (2) | 0.10735 (18) | 0.0696 (12) | |
| H5A | -0.0122 | -0.0279 | 0.1086 | 0.104* | |
| H5B | -0.1147 | -0.0566 | 0.0709 | 0.104* | |
| H5C | -0.1179 | -0.0636 | 0.1401 | 0.104* | |
| C6 | -0.2708 (3) | 0.0443 (3) | 0.10853 (16) | 0.0620 (11) | |
| H6A | -0.3049 | 0.0966 | 0.1106 | 0.093* | |
| H6B | -0.2959 | 0.0120 | 0.1412 | 0.093* | |
| H6C | -0.2914 | 0.0187 | 0.0720 | 0.093* | |
| C7 | -0.0953 (3) | 0.1046 (3) | 0.06081 (16) | 0.0636 (11) | |
| H7A | -0.0148 | 0.1034 | 0.0657 | 0.076* | |
| H7B | -0.1110 | 0.0752 | 0.0246 | 0.076* | |
| C8 | -0.1256 (3) | 0.1924 (3) | 0.04815 (17) | 0.0746 (12) | |
| C9 | -0.1280 (13) | 0.2466 (7) | 0.1047 (4) | 0.082 (3) | 0.61 (3) |
| H9A | -0.1479 | 0.3007 | 0.0938 | 0.124* | 0.61 (3) |

supplementary materials

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|------|--------------|---------------|--------------|-------------|-----------|
| H9B | -0.0555 | 0.2467 | 0.1229 | 0.124* | 0.61 (3) |
| H9C | -0.1820 | 0.2258 | 0.1321 | 0.124* | 0.61 (3) |
| C10 | -0.0428 (10) | 0.2347 (12) | 0.0057 (7) | 0.105 (4) | 0.61 (3) |
| H10A | -0.0659 | 0.2895 | -0.0003 | 0.157* | 0.61 (3) |
| H10B | -0.0419 | 0.2069 | -0.0316 | 0.157* | 0.61 (3) |
| H10C | 0.0306 | 0.2339 | 0.0227 | 0.157* | 0.61 (3) |
| C11 | -0.2464 (7) | 0.1932 (10) | 0.0228 (6) | 0.068 (3) | 0.61 (3) |
| H11A | -0.2684 | 0.2478 | 0.0147 | 0.103* | 0.61 (3) |
| H11B | -0.2966 | 0.1701 | 0.0513 | 0.103* | 0.61 (3) |
| H11C | -0.2487 | 0.1623 | -0.0131 | 0.103* | 0.61 (3) |
| C9' | -0.079 (3) | 0.2461 (13) | 0.0978 (9) | 0.105 (6) | 0.39 (3) |
| H9'A | -0.0983 | 0.3013 | 0.0902 | 0.158* | 0.39 (3) |
| H9'B | 0.0010 | 0.2409 | 0.0991 | 0.158* | 0.39 (3) |
| H9'C | -0.1099 | 0.2296 | 0.1351 | 0.158* | 0.39 (3) |
| C10' | -0.0550 (16) | 0.2007 (15) | -0.0096 (7) | 0.080 (5) | 0.39 (3) |
| H10D | -0.0634 | 0.2541 | -0.0254 | 0.119* | 0.39 (3) |
| H10E | -0.0803 | 0.1620 | -0.0382 | 0.119* | 0.39 (3) |
| H10F | 0.0223 | 0.1910 | -0.0007 | 0.119* | 0.39 (3) |
| C11' | -0.2416 (12) | 0.2242 (16) | 0.0303 (12) | 0.082 (6) | 0.39 (3) |
| H11D | -0.2384 | 0.2817 | 0.0256 | 0.123* | 0.39 (3) |
| H11E | -0.2948 | 0.2109 | 0.0605 | 0.123* | 0.39 (3) |
| H11F | -0.2638 | 0.1998 | -0.0065 | 0.123* | 0.39 (3) |
| C12 | 0.2972 (3) | 0.03718 (19) | 0.12475 (14) | 0.0412 (8) | |
| C13 | 0.4244 (3) | 0.0404 (2) | 0.11983 (18) | 0.0630 (11) | |
| H13A | 0.4483 | 0.0956 | 0.1179 | 0.095* | |
| H13B | 0.4478 | 0.0127 | 0.0846 | 0.095* | |
| H13C | 0.4572 | 0.0150 | 0.1539 | 0.095* | |
| C14 | 0.2474 (3) | 0.0762 (2) | 0.06931 (14) | 0.0589 (10) | |
| H14A | 0.2739 | 0.1306 | 0.0660 | 0.088* | |
| H14B | 0.1674 | 0.0763 | 0.0722 | 0.088* | |
| H14C | 0.2697 | 0.0461 | 0.0350 | 0.088* | |
| C15 | 0.2529 (3) | -0.05055 (19) | 0.12883 (15) | 0.0472 (9) | |
| H15A | 0.1725 | -0.0471 | 0.1243 | 0.057* | |
| H15B | 0.2806 | -0.0782 | 0.0939 | 0.057* | |
| C16 | 0.2736 (3) | -0.1075 (2) | 0.18095 (17) | 0.0587 (10) | |
| C17 | 0.1935 (7) | -0.0847 (4) | 0.2326 (3) | 0.089 (2) | 0.758 (9) |
| H17A | 0.2063 | -0.1200 | 0.2656 | 0.133* | 0.758 (9) |
| H17B | 0.1175 | -0.0899 | 0.2196 | 0.133* | 0.758 (9) |
| H17C | 0.2072 | -0.0299 | 0.2445 | 0.133* | 0.758 (9) |
| C18 | 0.3921 (5) | -0.1080 (4) | 0.2060 (3) | 0.092 (2) | 0.758 (9) |
| H18A | 0.4437 | -0.1229 | 0.1754 | 0.138* | 0.758 (9) |
| H18B | 0.3966 | -0.1461 | 0.2379 | 0.138* | 0.758 (9) |
| H18C | 0.4103 | -0.0551 | 0.2204 | 0.138* | 0.758 (9) |
| C19 | 0.2426 (7) | -0.1919 (3) | 0.1590 (4) | 0.088 (2) | 0.758 (9) |
| H19A | 0.2918 | -0.2072 | 0.1274 | 0.132* | 0.758 (9) |
| H19B | 0.1669 | -0.1917 | 0.1450 | 0.132* | 0.758 (9) |
| H19C | 0.2496 | -0.2297 | 0.1910 | 0.132* | 0.758 (9) |
| C17' | 0.277 (2) | -0.0737 (11) | 0.2437 (5) | 0.081 (6) | 0.242 (9) |
| H17D | 0.3526 | -0.0601 | 0.2538 | 0.122* | 0.242 (9) |

| | | | | | |
|------|-------------|--------------|-------------|-----------|-----------|
| H17E | 0.2495 | -0.1135 | 0.2709 | 0.122* | 0.242 (9) |
| H17F | 0.2312 | -0.0263 | 0.2459 | 0.122* | 0.242 (9) |
| C18' | 0.3859 (13) | -0.1499 (13) | 0.1679 (11) | 0.094 (6) | 0.242 (9) |
| H18D | 0.3987 | -0.1507 | 0.1259 | 0.141* | 0.242 (9) |
| H18E | 0.3833 | -0.2042 | 0.1826 | 0.141* | 0.242 (9) |
| H18F | 0.4454 | -0.1212 | 0.1871 | 0.141* | 0.242 (9) |
| C19' | 0.1904 (18) | -0.1786 (12) | 0.1821 (11) | 0.086 (7) | 0.242 (9) |
| H19D | 0.2102 | -0.2168 | 0.1519 | 0.129* | 0.242 (9) |
| H19E | 0.1163 | -0.1588 | 0.1748 | 0.129* | 0.242 (9) |
| H19F | 0.1929 | -0.2042 | 0.2201 | 0.129* | 0.242 (9) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|------|-------------|-------------|-------------|--------------|--------------|--------------|
| C11 | 0.0463 (5) | 0.0593 (6) | 0.0481 (5) | 0.0011 (4) | -0.0006 (4) | -0.0154 (4) |
| N1 | 0.0265 (14) | 0.0436 (15) | 0.0361 (14) | 0.0013 (12) | -0.0037 (13) | -0.0017 (12) |
| N3 | 0.0291 (15) | 0.0418 (15) | 0.0391 (15) | -0.0026 (12) | -0.0005 (12) | -0.0017 (13) |
| N2 | 0.0276 (14) | 0.0516 (17) | 0.0360 (15) | 0.0011 (12) | -0.0020 (12) | -0.0026 (13) |
| N4 | 0.0255 (14) | 0.0683 (19) | 0.0375 (15) | -0.0001 (13) | -0.0035 (13) | -0.0078 (14) |
| N5 | 0.0265 (14) | 0.0413 (15) | 0.0418 (16) | -0.0019 (12) | 0.0007 (12) | -0.0049 (12) |
| C1 | 0.0309 (17) | 0.0409 (18) | 0.0329 (17) | -0.0001 (15) | -0.0005 (15) | 0.0063 (15) |
| C2 | 0.0345 (18) | 0.0383 (18) | 0.0312 (16) | -0.0006 (16) | 0.0012 (16) | 0.0001 (14) |
| C3 | 0.0318 (18) | 0.0301 (17) | 0.0364 (18) | -0.0003 (14) | -0.0004 (15) | 0.0055 (15) |
| C4 | 0.0330 (19) | 0.071 (3) | 0.044 (2) | 0.0026 (18) | -0.0087 (16) | -0.0126 (18) |
| C5 | 0.053 (3) | 0.076 (3) | 0.080 (3) | 0.007 (2) | -0.008 (2) | -0.031 (2) |
| C6 | 0.035 (2) | 0.096 (3) | 0.055 (2) | -0.002 (2) | -0.0122 (18) | -0.018 (2) |
| C7 | 0.039 (2) | 0.108 (3) | 0.044 (2) | 0.007 (2) | -0.0056 (17) | -0.002 (2) |
| C8 | 0.061 (3) | 0.106 (4) | 0.056 (3) | -0.002 (3) | -0.007 (2) | 0.026 (3) |
| C9 | 0.070 (7) | 0.066 (5) | 0.111 (6) | -0.007 (5) | -0.021 (5) | 0.023 (4) |
| C10 | 0.087 (6) | 0.120 (8) | 0.107 (7) | -0.013 (6) | 0.014 (5) | 0.035 (6) |
| C11 | 0.063 (5) | 0.075 (7) | 0.067 (5) | 0.024 (4) | -0.011 (4) | 0.009 (5) |
| C9' | 0.110 (10) | 0.092 (8) | 0.113 (8) | -0.014 (8) | -0.013 (8) | 0.005 (7) |
| C10' | 0.068 (7) | 0.085 (9) | 0.086 (8) | -0.007 (6) | 0.005 (6) | 0.029 (6) |
| C11' | 0.074 (7) | 0.084 (10) | 0.088 (8) | 0.007 (6) | 0.004 (6) | 0.006 (8) |
| C12 | 0.0330 (18) | 0.0426 (19) | 0.048 (2) | 0.0004 (15) | 0.0044 (16) | -0.0034 (16) |
| C13 | 0.039 (2) | 0.071 (3) | 0.080 (3) | -0.004 (2) | 0.022 (2) | -0.023 (2) |
| C14 | 0.069 (3) | 0.067 (3) | 0.041 (2) | -0.001 (2) | 0.0097 (19) | 0.0026 (19) |
| C15 | 0.0397 (19) | 0.044 (2) | 0.058 (2) | 0.0040 (16) | -0.0058 (17) | -0.0059 (17) |
| C16 | 0.063 (3) | 0.043 (2) | 0.071 (3) | 0.0043 (19) | -0.010 (2) | 0.0045 (19) |
| C17 | 0.115 (6) | 0.074 (4) | 0.077 (4) | 0.009 (4) | 0.017 (4) | 0.035 (3) |
| C18 | 0.087 (4) | 0.092 (5) | 0.098 (5) | 0.010 (4) | -0.039 (4) | 0.018 (4) |
| C19 | 0.113 (6) | 0.052 (3) | 0.100 (5) | -0.013 (4) | -0.011 (4) | -0.001 (3) |
| C17' | 0.096 (10) | 0.074 (8) | 0.074 (8) | 0.004 (8) | -0.009 (8) | 0.019 (7) |
| C18' | 0.090 (9) | 0.087 (9) | 0.105 (10) | 0.007 (8) | -0.013 (8) | 0.013 (8) |
| C19' | 0.098 (10) | 0.068 (9) | 0.093 (10) | -0.014 (8) | 0.004 (8) | -0.008 (7) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|--------|-----------|-----------|--------|
| C11—C2 | 1.742 (3) | C10'—H10D | 0.9600 |
|--------|-----------|-----------|--------|

supplementary materials

| | | | |
|-----------|-----------|----------------|------------|
| N1—C3 | 1.341 (4) | C10'—H10E | 0.9600 |
| N1—C1 | 1.349 (4) | C10'—H10F | 0.9600 |
| N3—C2 | 1.311 (3) | C11'—H11D | 0.9600 |
| N3—C3 | 1.371 (4) | C11'—H11E | 0.9600 |
| N2—C2 | 1.315 (4) | C11'—H11F | 0.9600 |
| N2—C1 | 1.366 (4) | C12—C13 | 1.528 (4) |
| N4—C1 | 1.334 (4) | C12—C14 | 1.530 (5) |
| N4—C4 | 1.478 (4) | C12—C15 | 1.551 (5) |
| N4—H4 | 0.8600 | C13—H13A | 0.9600 |
| N5—C3 | 1.333 (4) | C13—H13B | 0.9600 |
| N5—C12 | 1.476 (4) | C13—H13C | 0.9600 |
| N5—H5 | 0.8600 | C14—H14A | 0.9600 |
| C4—C7 | 1.534 (5) | C14—H14B | 0.9600 |
| C4—C5 | 1.537 (5) | C14—H14C | 0.9600 |
| C4—C6 | 1.538 (4) | C15—C16 | 1.530 (5) |
| C5—H5A | 0.9600 | C15—H15A | 0.9700 |
| C5—H5B | 0.9600 | C15—H15B | 0.9700 |
| C5—H5C | 0.9600 | C16—C17' | 1.524 (9) |
| C6—H6A | 0.9600 | C16—C18 | 1.527 (6) |
| C6—H6B | 0.9600 | C16—C19 | 1.530 (6) |
| C6—H6C | 0.9600 | C16—C19' | 1.543 (10) |
| C7—C8 | 1.529 (6) | C16—C18' | 1.546 (9) |
| C7—H7A | 0.9700 | C16—C17 | 1.557 (6) |
| C7—H7B | 0.9700 | C17—H17A | 0.9600 |
| C8—C9' | 1.536 (9) | C17—H17B | 0.9600 |
| C8—C11' | 1.539 (9) | C17—H17C | 0.9600 |
| C8—C10 | 1.546 (8) | C18—H18A | 0.9600 |
| C8—C11 | 1.556 (7) | C18—H18B | 0.9600 |
| C8—C10' | 1.560 (9) | C18—H18C | 0.9600 |
| C8—C9 | 1.561 (8) | C19—H19A | 0.9600 |
| C9—H9A | 0.9600 | C19—H19B | 0.9600 |
| C9—H9B | 0.9600 | C19—H19C | 0.9600 |
| C9—H9C | 0.9600 | C17'—H17D | 0.9600 |
| C10—H10A | 0.9600 | C17'—H17E | 0.9600 |
| C10—H10B | 0.9600 | C17'—H17F | 0.9600 |
| C10—H10C | 0.9600 | C18'—H18D | 0.9600 |
| C11—H11A | 0.9600 | C18'—H18E | 0.9600 |
| C11—H11B | 0.9600 | C18'—H18F | 0.9600 |
| C11—H11C | 0.9600 | C19'—H19D | 0.9600 |
| C9'—H9'A | 0.9600 | C19'—H19E | 0.9600 |
| C9'—H9'B | 0.9600 | C19'—H19F | 0.9600 |
| C9'—H9'C | 0.9600 | | |
| C3—N1—C1 | 114.8 (2) | C8—C10'—H10F | 109.5 |
| C2—N3—C3 | 112.5 (2) | H10D—C10'—H10F | 109.5 |
| C2—N2—C1 | 112.6 (3) | H10E—C10'—H10F | 109.5 |
| C1—N4—C4 | 129.0 (3) | C8—C11'—H11D | 109.5 |
| C1—N4—H4 | 115.5 | C8—C11'—H11E | 109.5 |
| C4—N4—H4 | 115.5 | H11D—C11'—H11E | 109.5 |
| C3—N5—C12 | 129.2 (3) | C8—C11'—H11F | 109.5 |

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| C3—N5—H5 | 115.4 | H11D—C11'—H11F | 109.5 |
| C12—N5—H5 | 115.4 | H11E—C11'—H11F | 109.5 |
| N4—C1—N1 | 120.5 (3) | N5—C12—C13 | 106.2 (3) |
| N4—C1—N2 | 114.5 (3) | N5—C12—C14 | 110.1 (3) |
| N1—C1—N2 | 125.0 (3) | C13—C12—C14 | 108.3 (3) |
| N3—C2—N2 | 130.0 (3) | N5—C12—C15 | 111.8 (3) |
| N3—C2—C11 | 115.3 (2) | C13—C12—C15 | 112.2 (3) |
| N2—C2—C11 | 114.8 (2) | C14—C12—C15 | 108.2 (3) |
| N5—C3—N1 | 120.7 (3) | C12—C13—H13A | 109.5 |
| N5—C3—N3 | 114.2 (3) | C12—C13—H13B | 109.5 |
| N1—C3—N3 | 125.1 (3) | H13A—C13—H13B | 109.5 |
| N4—C4—C7 | 111.9 (3) | C12—C13—H13C | 109.5 |
| N4—C4—C5 | 110.3 (3) | H13A—C13—H13C | 109.5 |
| C7—C4—C5 | 108.5 (3) | H13B—C13—H13C | 109.5 |
| N4—C4—C6 | 105.1 (3) | C12—C14—H14A | 109.5 |
| C7—C4—C6 | 113.2 (3) | C12—C14—H14B | 109.5 |
| C5—C4—C6 | 107.7 (3) | H14A—C14—H14B | 109.5 |
| C4—C5—H5A | 109.5 | C12—C14—H14C | 109.5 |
| C4—C5—H5B | 109.5 | H14A—C14—H14C | 109.5 |
| H5A—C5—H5B | 109.5 | H14B—C14—H14C | 109.5 |
| C4—C5—H5C | 109.5 | C16—C15—C12 | 124.7 (3) |
| H5A—C5—H5C | 109.5 | C16—C15—H15A | 106.1 |
| H5B—C5—H5C | 109.5 | C12—C15—H15A | 106.1 |
| C4—C6—H6A | 109.5 | C16—C15—H15B | 106.1 |
| C4—C6—H6B | 109.5 | C12—C15—H15B | 106.1 |
| H6A—C6—H6B | 109.5 | H15A—C15—H15B | 106.3 |
| C4—C6—H6C | 109.5 | C17'—C16—C18 | 68.4 (10) |
| H6A—C6—H6C | 109.5 | C17'—C16—C15 | 119.5 (8) |
| H6B—C6—H6C | 109.5 | C18—C16—C15 | 116.0 (4) |
| C8—C7—C4 | 125.4 (3) | C17'—C16—C19 | 130.1 (9) |
| C8—C7—H7A | 106.0 | C18—C16—C19 | 110.0 (4) |
| C4—C7—H7A | 106.0 | C15—C16—C19 | 106.0 (4) |
| C8—C7—H7B | 106.0 | C17'—C16—C19' | 106.4 (13) |
| C4—C7—H7B | 106.0 | C18—C16—C19' | 126.2 (10) |
| H7A—C7—H7B | 106.3 | C15—C16—C19' | 112.3 (10) |
| C7—C8—C9' | 109.2 (10) | C19—C16—C19' | 31.8 (9) |
| C7—C8—C11' | 126.1 (11) | C17'—C16—C18' | 108.7 (13) |
| C9'—C8—C11' | 108.8 (11) | C18—C16—C18' | 42.3 (9) |
| C7—C8—C10 | 113.3 (8) | C15—C16—C18' | 106.0 (9) |
| C9'—C8—C10 | 87.5 (9) | C19—C16—C18' | 74.5 (10) |
| C11'—C8—C10 | 105.1 (11) | C19'—C16—C18' | 102.6 (13) |
| C7—C8—C11 | 107.3 (7) | C17'—C16—C17 | 39.7 (9) |
| C9'—C8—C11 | 127.0 (11) | C18—C16—C17 | 107.2 (5) |
| C11'—C8—C11 | 20.2 (10) | C15—C16—C17 | 109.0 (3) |
| C10—C8—C11 | 111.4 (9) | C19—C16—C17 | 108.4 (5) |
| C7—C8—C10' | 96.4 (9) | C19'—C16—C17 | 77.0 (10) |
| C9'—C8—C10' | 111.2 (10) | C18'—C16—C17 | 142.2 (9) |
| C11'—C8—C10' | 103.9 (12) | C16—C17—H17A | 109.5 |
| C10—C8—C10' | 25.2 (8) | C16—C17—H17B | 109.5 |

supplementary materials

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| C11—C8—C10' | 101.3 (10) | H17A—C17—H17B | 109.5 |
| C7—C8—C9 | 113.6 (5) | C16—C17—H17C | 109.5 |
| C9'—C8—C9 | 22.7 (10) | H17A—C17—H17C | 109.5 |
| C11'—C8—C9 | 90.1 (10) | H17B—C17—H17C | 109.5 |
| C10—C8—C9 | 104.9 (8) | C16—C18—H18A | 109.5 |
| C11—C8—C9 | 106.2 (7) | C16—C18—H18B | 109.5 |
| C10'—C8—C9 | 130.0 (9) | H18A—C18—H18B | 109.5 |
| C8—C9—H9A | 109.5 | C16—C18—H18C | 109.5 |
| C8—C9—H9B | 109.5 | H18A—C18—H18C | 109.5 |
| H9A—C9—H9B | 109.5 | H18B—C18—H18C | 109.5 |
| C8—C9—H9C | 109.5 | C16—C19—H19A | 109.5 |
| H9A—C9—H9C | 109.5 | C16—C19—H19B | 109.5 |
| H9B—C9—H9C | 109.5 | H19A—C19—H19B | 109.5 |
| C8—C10—H10A | 109.5 | C16—C19—H19C | 109.5 |
| C8—C10—H10B | 109.5 | H19A—C19—H19C | 109.5 |
| H10A—C10—H10B | 109.5 | H19B—C19—H19C | 109.5 |
| C8—C10—H10C | 109.5 | C16—C17'—H17D | 109.5 |
| H10A—C10—H10C | 109.5 | C16—C17'—H17E | 109.5 |
| H10B—C10—H10C | 109.5 | H17D—C17'—H17E | 109.5 |
| C8—C11—H11A | 109.5 | C16—C17'—H17F | 109.5 |
| C8—C11—H11B | 109.5 | H17D—C17'—H17F | 109.5 |
| H11A—C11—H11B | 109.5 | H17E—C17'—H17F | 109.5 |
| C8—C11—H11C | 109.5 | C16—C18'—H18D | 109.5 |
| H11A—C11—H11C | 109.5 | C16—C18'—H18E | 109.5 |
| H11B—C11—H11C | 109.5 | H18D—C18'—H18E | 109.5 |
| C8—C9'—H9'A | 109.5 | C16—C18'—H18F | 109.5 |
| C8—C9'—H9'B | 109.5 | H18D—C18'—H18F | 109.5 |
| H9'A—C9'—H9'B | 109.5 | H18E—C18'—H18F | 109.5 |
| C8—C9'—H9'C | 109.5 | C16—C19'—H19D | 109.5 |
| H9'A—C9'—H9'C | 109.5 | C16—C19'—H19E | 109.5 |
| H9'B—C9'—H9'C | 109.5 | H19D—C19'—H19E | 109.5 |
| C8—C10'—H10D | 109.5 | C16—C19'—H19F | 109.5 |
| C8—C10'—H10E | 109.5 | H19D—C19'—H19F | 109.5 |
| H10D—C10'—H10E | 109.5 | H19E—C19'—H19F | 109.5 |
| C4—N4—C1—N1 | -6.4 (5) | C5—C4—C7—C8 | 178.9 (3) |
| C4—N4—C1—N2 | 172.2 (3) | C6—C4—C7—C8 | 59.5 (5) |
| C3—N1—C1—N4 | 179.7 (3) | C4—C7—C8—C9' | 69.0 (13) |
| C3—N1—C1—N2 | 1.2 (4) | C4—C7—C8—C11' | -63.6 (14) |
| C2—N2—C1—N4 | -177.6 (3) | C4—C7—C8—C10 | 164.7 (9) |
| C2—N2—C1—N1 | 1.0 (4) | C4—C7—C8—C11 | -71.9 (7) |
| C3—N3—C2—N2 | 3.2 (4) | C4—C7—C8—C10' | -175.9 (10) |
| C3—N3—C2—C11 | -175.90 (19) | C4—C7—C8—C9 | 45.1 (8) |
| C1—N2—C2—N3 | -3.4 (5) | C3—N5—C12—C13 | 175.0 (3) |
| C1—N2—C2—C11 | 175.6 (2) | C3—N5—C12—C14 | 58.0 (4) |
| C12—N5—C3—N1 | -2.9 (4) | C3—N5—C12—C15 | -62.2 (4) |
| C12—N5—C3—N3 | 177.6 (3) | N5—C12—C15—C16 | -52.4 (4) |
| C1—N1—C3—N5 | 179.0 (3) | C13—C12—C15—C16 | 66.8 (4) |
| C1—N1—C3—N3 | -1.5 (4) | C14—C12—C15—C16 | -173.7 (3) |
| C2—N3—C3—N5 | 179.1 (2) | C12—C15—C16—C17' | 35.9 (13) |

| | | | |
|-------------|-----------|------------------|------------|
| C2—N3—C3—N1 | -0.4 (4) | C12—C15—C16—C18 | -42.9 (6) |
| C1—N4—C4—C7 | -59.6 (4) | C12—C15—C16—C19 | -165.3 (5) |
| C1—N4—C4—C5 | 61.4 (4) | C12—C15—C16—C19' | 161.6 (11) |
| C1—N4—C4—C6 | 177.2 (3) | C12—C15—C16—C18' | -87.2 (11) |
| N4—C4—C7—C8 | -59.1 (4) | C12—C15—C16—C17 | 78.2 (5) |

Hydrogen-bond geometry (\AA , $^\circ$)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|---------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| N4—H4 \cdots N3 ⁱ | 0.86 | 2.39 | 3.245 (4) | 174 |
| N5—H5 \cdots N2 ⁱⁱ | 0.86 | 2.29 | 3.147 (3) | 179 |

Symmetry codes: (i) $x-1/2, y, -z+1/2$; (ii) $x+1/2, y, -z+1/2$.

Fig. 1

